## THE DIRECTED ORTHO METALATION CONNECTION TO ARYL-ARYL CROSS COUPLING. A GENERAL REGIOSPECIFIC SYNTHESIS OF PHENANTHROLS

J.-m. Fu, M.J. Sharp, and V. Snieckus<sup>\*</sup>

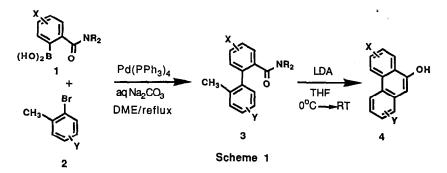
Guelph-Waterloo Centre for Graduate Work in Chemistry

University of Waterloo, Waterloo, Ontario, Canada N2L 3G1

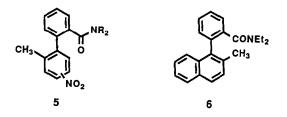
Summary: A general directed metalation-based cross coupling synthesis of phenanthrols 4 has been developed (Scheme 1); reactions of derived triflates and carbamates 10 lead to a variety of substituted phenanthrenes  $7 \rightarrow 12$  (Scheme 2).

The phenanthrenes constitute a significant class of fossil fuel-derived aromatic hydrocarbons,<sup>1</sup> an expanding group of natural products,<sup>2</sup> and synthons for several major classes of alkaloids.<sup>3</sup> Synthetic approaches to polysubstitued phenanthrenes are based on classical Pschorr,<sup>4</sup> Ullmann<sup>5</sup> and, more recently, Mallory photocyclization<sup>6</sup> methodologies. Most methods are compromised in efficacy and brevity by their dependence on non-regioselective synthetic protocols for substituted benzene derivatives.

We report a new general and efficient methodology for phenanthrenes based on the vinylogous acidity of the 2'-methyl group in the conversion  $3 \rightarrow 4$  (Scheme 1) and on the ready availability of the precursors 3 by the recently developed cross coupling regimen of arylboronic acids with aryl bromides.<sup>7</sup> Since benzamide <u>o</u>-boronic acids 1 are conveniently and rapidly obtained by the regiospecific directed ortho metalation strategy, <sup>7a,8</sup> the new phenanthrene synthesis promises overall advantage compared to classical methods.<sup>4,5,6</sup>



Treatment of the biphenyl diethyl amide 3 (X = Y = H, R = Et)<sup>9</sup> with LDA (2.5 equiv/THF/ 0°C  $\rightarrow$  rt, 3 h) gave 9-phenanthrol (Table, entry 1) in high yield. The corresponding diisopropyl amide 3 (X = Y = H, R = i-Pr) afforded 9-phenanthrol in even higher yield thus suggesting either a lack of a steric effect in the carbinolamine-forming step or a driving force for expulsion of the bulky  $N(i-Pr)_2$  leading to a stable phenolate. A number of substituted biphenyls<sup>9</sup> likewise led in good to excellent yields to phenanthrols bearing Cl, OMe, and CONEt<sub>2</sub> substituents (entries  $3 \rightarrow 7$ ). The facile synthesis of the known<sup>10</sup> 5-hydroxybenz[a]anthracene (entry 8) suggests applicability of this method to more highly condensed polycyclic aromatics. Although nitrobiphenyls 5 were readily prepared by the cross coupling regimen,<sup>9</sup> their cyclization under LDA (up to 4 equiv LDA/THF/reflux) or other basic (e.g. NaH) conditions was unsuccessful.<sup>11</sup> The naphthylphenyl coupled product 6 also failed to undergo cyclization presumably due to a developing peri hydrogen-phenyl interaction.



To demonstrate accessibility of the parent hydrocarbons, two phenanthrols were transformed into triflates 10a and 10b (Scheme 2) and thence into the corresponding phenanthrenes 7a and 7b using emerging methodology.<sup>12</sup> Significantly, this overall route is favorably competitive with previous published procedures, e.g. for the synthesis of methoxy phenanthrenes corresponding to phenanthrols of entries 3 and  $7^{13}$ . To provide additional scope to the new phenanthrene synthesis, the prototype 9-O-carbamate 10c was metalated (t-BuLi/THF/-78°C) and converted into either substituted (9a, 92%; 9b, 88%) or anionic Fries rearranged (12, 81%) products.<sup>14</sup> Furthermore, the boronic acid 9c, obtained from 9b,<sup>7b</sup> was cross coupled with 1-bromo-4-nitrobenzene to give 8 (80%) and the parent triflate 10a led, upon treatment with o-stannylated benzamide<sup>15</sup> under Stille's conditions,<sup>16</sup> to biaryl 11 (76%).

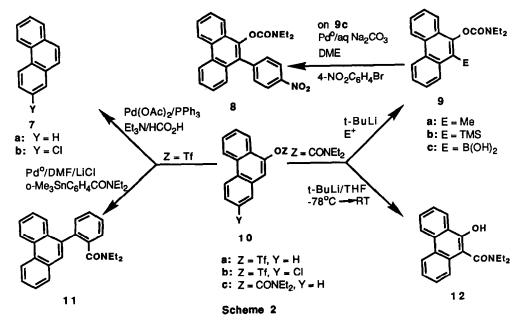


Table. Synthesis of Phenanthrols

Entry	Biphenyl	Phenanthrol	Yield, % <sup>a</sup>	mp °C
		HO		
1 2	С́Н <sub>3</sub> R = Et R = i-Pr		92 98	151-152 <sup>6</sup> (PhH)
3	Et <sub>2</sub> N MeO CH <sub>3</sub>		92	133-134 (cyclohexane)
4			78	217-219 (EtOAc)
5			93	1 6 4 - 1 6 7 (hexane - CH <sub>2</sub> Cl <sub>2</sub> )
6			96	119.5-120.5 <sup>c</sup> (hexane)
7			92	75-77 <sup>d</sup> (hexane- CH <sub>2</sub> Cl <sub>2</sub> )
8			90	124-127 <sup>d,e</sup> (hexane- CH <sub>2</sub> Cl <sub>2</sub> )

<sup>a</sup>Represent yields of products purified by column chromatography. Most phenanthrols were somewhat unstable under ambient conditions; <sup>b</sup>Lit. mp 149-152°C (Hunsberger, I. M.; Ketcham, R.; Gutowsky, H. S. J. Am. Chem. Soc. 1952, 74, 4839); <sup>c</sup>Mp of triflate; <sup>d</sup>Mp of acetate; <sup>e</sup>Lit.<sup>10</sup>mp 127-128°C.

In summary, a general and efficient methodology for the regiospecific construction of polysubstituted phenanthrenes from biphenyls has been devised. Its dependence on the powerful directed ortho metalation and cross coupling strategies<sup>7,8</sup> provide it broad scope and versatility while further transformations of resulting phenanthrols offer new opportunities in aromatic chemistry.<sup>17,18</sup>

## **References and Footnotes**

- 1. Dipple, A.; Moschel, R.C.; Bigger, C.A.H. In <u>Chemical Carcinogens</u>, 2nd ed.; Searle, C.E., Ed.; ACS Monograph 182; American Chemical Society: Washington, D.C, 1984; Vol. 1, p 41.
- Majumder, P.L.; Kar, A. <u>Phytochemistry</u> 1987, 26, 1127; Carvalho, C.F.; Sargent, M.V.; Stanojevic, E. <u>Austr.</u> J. Chem. 1984, 37, 2111; Boger, D.C.; Mullican, M.D. J. Org. Chem. 1984, 49, 4045.
- Shamma, M. <u>The Isoquinoline Alkaloids</u>, Academic Press: New York, 1972, Chapter 14; Bick, I.R.C.; Sinchai, W. <u>The Alkaloids</u> 1981, 19, 193; Kametani, T.; Honda, T. <u>ibid</u>, 1985, 24, 153.
- 4. Floyd, A.J.; Dyke, S.F.; Ward, S.E. Chem. Rev. 1976, 76, 509.
- 5. Ames, D.E.; Opalko, A. <u>Tetrahedron</u> 1984, <u>40</u>, 1919.
- 6. Mallory, F.B.; Mallory, C. Org. React. 1984, 30, 1.
- a) Sharp, M.J.; Snieckus V. <u>Tetrahedron Lett.</u> 1985, <u>26</u>, 5997; b) Sharp, M.J.; Cheng, W.; Snieckus, V. <u>ibid.</u> 1987, <u>28</u>, 5093; c) Cheng, W.; Snieckus, V. <u>ibid.</u> 1987, <u>28</u>, 5097.
- 8. Snieckus, V. Bull. Soc. Chim. France (II), 1988, 67.
- 9. Prepared according to the procedure  $^{7a,7b}$  in 67-98% yields with the exception of biphenyl of entry 7 (23%).
- 10. Harvey, R.G.; Goh, S.H.; Cortez, C. J. Am. Chem. Soc. 1975, 97, 3468.
- 11. In all cases, starting material was recovered. SET reactions may be operating, see Houge-Frydrych, C.S.V.; Motherwell, W.B.; O'Shea, D.M. J.C.S., Chem. Commun. 1987, 1819 and refs. therein.
- Chen, Q.Y.; He, Y.B.; Yang, Z.Y. J. Chem. Soc., Chem. Commun. 1986, 1452; Cacchi, S.; Ciattini, P.G.; Morere, E.; Ortar, G. <u>Tetrahedron Lett.</u> 1986, 27, 5541; Peterson, G.A.; Kunng, F.A.; McCallum, J.S.; Wulff, W.D. <u>ibid.</u> 1987, 28, 1381.
- Abdel-Wahhab, S.M.; El-Assal, L.S.; Ramses, N.; Shehab, A.H. J. Chem. Soc. (C), 1968, 863; Giles, R.G.F.; Sargent, M.V. J. Chem. Soc., Chem. Commun. 1974, 215; Brown, C.; Sikkel, B.J.; Carvalho, C.F.; Sargent, M.V. J. Chem. Soc., Perkin Trans. I 1982, 3007.
- 14. Sibi, M.P.; Snieckus, V. J. Org. Chem. 1983, 48, 1935.
- 15. Reed, J.N. Ph.D. Thesis, University of Waterloo, 1985.
- 16. Echavarren, A.M.; Stille, J.K. J. Am. Chem. Soc. 1987, 109, 5478.
- 17. All new compounds show analytical and spectral (IR, NMR, MS) data fully consistent with the given structures.
- 18. Financial assistance from NSERC Canada, Merck Frosst Canada, and Ministry of the Environment (Ontario) in support of our synthetic programs is gratefully acknowledged.

(Received in USA 14 July 1988)